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Identification and Analysis of PM_{2.5} WSI₈ and Trace Gases during Winter Season at Industrial Zone of Tatanagar

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ABSTRACT

This study has been conducted from Dec 2014 to March 2015 and data of fine particle and trace gases $(SO_2 \text{ and } NO_2)$ were collected. The high volume sampler was used for collection of $PM_{2.5}$ and trace gases $(SO_2 \& NO_2)$ were collected with impinger equipped with PM sampler. The average winter season $PM_{2.5}$ mass concentrations from Industrial (A-1), Traffic (A-2), Academic (A-3) and Rural (A-4) sampling sites of Adityapur industrial zone, Tata nagar are $178\pm68 \ \mu g/m^3$, $102\pm56 \ \mu g/m^3$, $174.5\pm84 \ \mu g/m^3$ and $88.25\pm37 \ \mu g/m^3$ respectively. Water soluble ionic species ions $(Na^+, NH_4^+, K^+, Mg^{2+}, C\Gamma, NO_3^- \text{ and } SO_4^{2-})$ were analyzed with ion chromatography. The three most abundant ions were SO_4^{2-} , NO^{3-} , and NH_4^+ with average concentrations of 6.41 ± 1.5 , 2.40 ± 0.30 and $2.96\pm0.62 \ \mu g/m^3$, respectively. The trace gaseous present in study area having average concentration of SO_2 and NO_2 were much lower than the standards guideline values of the WHO. The evolution patterns clearly show the winter season of fine $PM_{2.5}$ with impaction low temperature. An air-mass pathway traced using HYSPLIT model over the study area illustrates the direction and dispersion of pollutants.

Keywords: PM_{2.5} Trace gases, Water soluble ions, WHO. HYSPLIT model.

INTRODUCTION

The $PM_{2.5}$ study of atmospheric aerosols is gaining importance in scientific world as lot of studies have shown their association with health-related problems [1] and climate changes [2]. As from the previous literature studies it has been well established the deleterious impact of ambient particulate matter (PM) on human, animal as well as plants health and the World Health Organization (WHO) from time to time publishes suggested air quality guidelines for mass concentrations of $PM_{2.5}$ size fractions (particles with aerodynamic diameters smaller than 2.5 µm). Due to increasing anthropogenic activities such as increasing vehicular traffic, biomass burning, increased industrial related activities, forest burning, fuel wood and cow dung burning for cooking may alter the pollutants level in the region.

A large portion of aerosol particles comprise of water-soluble ions (WSI) which plays an important role in the atmospheric chemistry. The factors which influence the mass concentrations of aerosol particles are meteorological factors, geographic conditions and particle emissions sources. The chemical characteristic

properties of aerosols are due to their water soluble components, e.g. potassium, magnesium, calcium, sodium, ammonium, chloride, nitrate, sulphate [3] and metals [4,5] that originate from the different sources through a series of complex mechanisms [4].

Meteorology plays an important role in ambient distributions of PM in atmosphere. The variations of meteorological variables for example temperature, wind speed and direction, humidity, mixing height and precipitation modulate the air quality which plays a major role in determining the levels, transport and diffusion of pollutants [2]. The inflowing of pollutants from the ground surface, their residence time and the formation of secondary pollutants in the atmosphere are not only controlled by rate of emission of the reactants from the source, but also by wind speed, turbulence level, air temperature and precipitation [6]. Thus, it is very important to understand and to study the physical phenomena like meteorological factors to determine the pollutant levels and their relationship with meteorological parameters.

MATERIALS AND METHODS

Study area: Tata Nagar is 86°04' to 86°54' east Longitude and 22°12' to 23°01' North latitude. It has an average elevation of 40 metres (131 feet). The city is situated 1400 km from New Delhi and a little less than 300 km north-west of Kolkata on the(NH-33) and Eastern Railway. The chief rivers are the swarnrekha and Kharkai. Tata nagar is located in the southern east region of Jharkhand. The city is located in the Chhota Nagpur plateau and covers an area of 3533 km². Sampling site is close to state highway Jamshedpur kandra road, Godowns of Food Corporation of India, hotels, bakeries and Adityapur railway station also exist in the vicinity of sampling area. Tata nagar is one of the major industrial centres of Eastern India. It houses companies like Tata Steel, TCE, Lafarge, Tata Motors, BOC Gases, Tata Power, Cement, Telcon, Praxair, TCS, Timken, TRF, Tinplate and many more. It homes to one of the largest industrial zones of India known as Adityapur which houses more than 1,200 small and medium scale industries and has a SEZ named AIDA in the Adityapur. The details of sampling sites for the collection of PM_{2.5} are shown in table 1.

Sampling location	oling location Site Description of		Type of sources		
	code				
Adityapur	A-1	Industrial and Traffic area	Industries, road dust, vehicles,		
Industrial (SEZ)			Domestic cooking, DG sets, garbage		
Area			burning.		
Shire Punjab	A-2	Commercial and Traffic	DG sets, vehicles, road dust, garbage		
Adityapur		cum residential area	burning, Domestic cooking,		
			restaurants.		
Nit Campus	A-3	Institutional cum residential	Domestic cooking, light vehicle.		
		area			
Burgidih	A-4	Rural Residential area	Domestic cooking, vehicles, road		
			dust.		

 Table 1: Sampling sites description of Adityapur industrial zone (Tatanagar).

Aim and Objective of this research work:

- * The assessment of concentration and the ambient air quality with respect to $PM_{2.5}WSI$, SO_2 and NO_2 .
- * Use of backward trajectory for the study of pattern of pollutants over sampling periods at Adityapur industrial zone, Tata nagar.
- * To create a database for further analysis.

Meteorological details of the site: Metrology plays crucial role in the $PM_{2.5}$ studies. There is a strong reaction between winter season & change in air quality level [7]. The Tata nagar has a temperate type climate. The temperature during study period varies between 8–41°C. The minimum temperature recorded

in Dec and Jan is 8° C. The climate of Tata nagar is marked by south-west monsoon. The details metrological data are shown in table 2.

during Dec 2014 to March 2015									
Month→	Dec	Jan	Feb	March					
parameter↓	(2014)	(2015)	(2015)	(2015)					
Temp (°C)									
(Max–Min)	(30, 0), 10	(28, 8) 10	(30, 0), 21	(39-9) 21.4					
Average	(30-9) 19	(20-0) 19	(39-9) 21						
Pressure (hpa)									
(Max–Min)	(1025-1009)	(1021-1012)	(1024-1007)	(1020-1003)					
Average	1017	1016	1015	1021					
Precipitation (mm)									
(Max–Min)		(1.0) 0.1	(17.0) 0.0	(10, 0), 0, 6					
Average	(0-0) 0	(1-0) 0.1	(17-0) 0.9	(10-0) 0.0					
Humidity (%)									
(Max –Min)	(34.08) 68	(31,00),71	(33,01) 66	(31.08).75					
Average	(34-98) 08	(31-90) /1	(33-91)00	(31-98) 75					
Wind speed (km/h)									
(Max –Min)	(37.0) 2	(11.0) 2	(150, 0) 3	(11.0) 2					
Average	(37-0) 2	(11-0) 2	(159-0) 5	(11-0) 2					
Wind direction	W-N	W-N	W-N	W-N					

 Table 2: The Ranges and Average meteorological parameters at sampling sites

 during Dec 2014 to March 2015

Sample Collection: Samples of PM_{2.5}, and gaseous were collected simultaneously from Dec 2014 to March 2015 in Tata nagar. A total of 18 samplers in including 2 field sample were deployed for our study among four categories of sites: Industrial area (Adityapur, A-1), Traffic (Shire Punjab Adityapur, A-2), Academic (Nit Campus, A-3) and Rural area (Burgidih, A-4) (>100 m from the nearest building) (Fig. 1). Samples were collected (every Monday on weekly basis; 4 to 5 samples in a month) on quartz filter fibre (that were prebaked at 550 °C at least 5h before desiccated and sample collection) by using Particle Sampler (APM460NL, Make: M/s. Envirotech, India) at 10m height (above ground level). Through Whatman Quartz Microfiber filter (QMA; size: 20×25 cm²) Ambient air was passed at a flow rate of 1.12 m³ min/m (accuracy 64%) for 8 h during the sampling period (9000–1700h). The QM-A filters were weighed before and after the sampling during the experiment in order to determine the mass of the PM_{2.5} collected. The amount of $PM_{2.5}$ (µg/m) was calculated on the basis of the difference between initial and final weights of the QM-A filters measured by a microbalance (M/s. Mettle-Toledo) was determined by dividing the amount of total volume passed during the sampling. After the collection of samples, filters were stored under dry condition at -20 ^oC in the deep-freezer prior to analysis. Meteorological data such as wind speed, temperature, relative humidity, pressure, precipitation and wind direction were obtained from Weather Underground (http://www.underground.com/) and data of SO₂ and NO_x were obtained from the website http://www.gzepb.gov.cn/.



Fig 1: Location of sampling sites (A-1, A-2, A-3, and A-4).

Analysis of water-soluble ionic species: For the analysis of water soluble ions (WSI) in PM_{2.5} sample the filters were extracted by ultrasonic agitation for 90 min. The extract is filtered through Whatman filter paper and stored in polypropylene sample bottles (these bottles are dipped in 2% HNO₃ and then dipped in deionised distilled water overnight to eliminate any contamination on these bottles). The major WSI_s Concentrations of (F^- , CI^- , NO_3^- and $SO_4^{2^-}$) are determined by Ion Chromatograph (DIONEX-ICS-3000, USA) using an Ion Pac-AS11-HC analytical column (4x250 mm, Dionex, USA) with a guard column (IonPac AG11-HC, 4 x 50 mm, Dionex, USA), ASRS-300 4 mm anion micro-membrane suppressor, 20 mM NaOH (50% w/w) as effluent and triple-distilled water as a regenerator. Na⁺, K⁺, Ca²⁺ and Mg²⁺ are determined by using a separation column (IonPac CS17-HC, 4 x250 mm, Dionex, USA) with a guard column (IonPac CG17-HC, 4 x 50 mm, Dionex, USA), suppressor CSRS-300 (4 mm, Dionex, USA) and 5 mM MSA as eluent. The IC system is fitted with a 25 mL sample loop that is used to introduce the sample manually. All the standard solutions are filtered using 0.45 mm nylon membrane filters (Millipore) and degassed by ultra-sonication.

RESULTS AND DISCUSSION

Ambient air PM_{2.5} concentrations: PM_{2.5} samples were collected at A-1, A-2, A-3 and A-4 during winter season. At these sites, the average mass concentrations ranked in the order of A-1 ($178\pm68 \ \mu g/m^3$) > A-3 ($174\pm84\mu g/m^3$) > A-2 ($102\pm56 \ \mu g/m^3$) > A-4 ($88.25\pm37 \ \mu g/m^3$) sites fig 3. This may be attributed to increased emissions from vehicular, construction and industrial exhaust. Fig 2 compares the PM_{2.5} with different sampling sites of India. On comparison, the average mass concentrations of PM_{2.5} was found to be lower than reported at other cities of India: [8] Kanpur ($154 \ \mu g/m^3$), [9] Raipur ($185.9 \pm 66.9 \ \mu g/m^3$), and [10] Agra ($170.4 \pm 54.9 \ \mu g/m^3$) but found to be higher than reported at [11] Lucknow ($101 \ \mu g/m^3$;), [12] Mumbai ($42 \ \mu g/m^3$) and [13] Ahmedabad ($55.7 \pm 17 \ \mu g/m^3$). The average daily mass concentration of PM during the measurement period exceeded the 24 h National Ambient Air Quality Standard of India ($60 \ \mu g/m^3$ for PM_{2.5} NAAQS, 2009) and WHO ($25 \ \mu g/m^3$ for PM_{2.5} whilibdoc.who.int/hq2006/WHO_SDE_PHE_OEH_06.02_eng.pdf) 24–h guidelines 100% of the time, respectively.



Sites

Fig 2: Showing PM_{2.5} Concentration of different study



Fig 3: PM_{2.5} Conc of four different sampling sites in the month of December 14 – March 15.

WSI composition: WSI comprise a large part of aerosol particles and play an important role in the atmosphere. In this study, seven major water-soluble inorganic components were detected (Cl⁻, NO₃⁻, SO₄²⁻, Na⁺, NH₄⁺, K⁺, Mg₂⁺). From fig 4, also presents the mean water soluble ionic concentrations in A-1 site as follows:

 $SO_4^{2-}\!>NH_4^+>K^+>\!NO_3^->Cl^-\!>Na^+\!>\!Mg^{2+}$

The mean water soluble ionic concentrations in A-2 site as follows: $SO_4^{2-} > K^+ > NH_4^+ > NO_3^- > Cl^- > Na^+ > Mg^{2+}$

The mean water soluble ionic concentrations in A-3 site as follows: $SO_4^{2-} > NH_4^+ > K^+ > NO_3^- > Cl^- > Na^+ > Mg^{2+}$

The mean water soluble ionic concentrations in A-4 site as follows: $SO_4^{2-} > K^+ > NH_4^+ > NO_3^- > Cl^- > Na^+ > Mg^{2+}$

High concentrations in winter may probably be attributed to the enhanced emissions from heating sources and stagnant atmospheric conditions (low temperature, low wind speed, low mixing height). The ionic constituents (Cl^- , NO_3^- , SO_4^{-2} , NH_4^+ and K^+) showed high mass concentrations in winter. The variation in

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ionic constituents is due to the change in meteorological parameters and physico-chemical transfer processes arising in the atmosphere. A comparative study has been made and tabled in table 3.

Table 3: Comparative table of wish of present study with previous study									
L Site	Type	Cl	NO_3^-	SO_4^{2-}	NH_4^+	Na^+	\mathbf{K}^+	Mg^{2+}	Ref.
Present study	PM _{2.5}	2.1	2.40	6.41	2.96	2.05	2.9	1.9	Present study
Raipur (India)	PM _{2.5}	6.8	8.1	46.5	8.7	7.4	5.9	1.6	[23]
New Delhi (India)	PM _{2.5}	12.7	6.41	19.8	-	9.8	4.8	0.9	[22]
Mumbai (India)	PM _{2.5}	4.6	0.9	11.4	3.7	3.1	2.8	0.9	[21]
Kanpur (India)	PM _{2.5}	0.6	5.7	13.5	5.2	0.2	2.4	0.04	[20]
Taichung (Taiwan)	PM _{2.5}	1.5	10.5	12.6	7.8	0.2	0.5	0.7	[19]
Shanghai (China)	PM _{2.5}	3.0	6.2	10.4	3.7	0.5	0.6	0.2	[18]
~	= = - 2.5		•			0.0	0.0	•	[-•]



Na+ Mg2+ K+ **NH4**+ Cl-**NO3-**SO4--Fig 4: WSI Conc of four different sampling sites in the month of December 14 – March 15

Gaseous (SO₂ and NO₂): As per the National Ambient Air Quality Standards (NAAQS), the limit for NO₂ and SO₂ are 80ug/m^3 is respectively. The concentration of NO₂ and SO₂ are 38.25 ± 4.55 and 52.18 ± 4.85 μ g/m³, figs 5, 6 respectively. The NO₂ and SO₂ levels were found to be below the permissible limit as prescribed by NAAQS. The conc of NO_2 are higher than SO_2 at the sampling site on all days of sampling. High levels of SO₂ are particularly dangerous in the presence of particulate matter because it slowly adhere on PM_{2.5} atmospheric particles and can be transported very deep into lungs[14] and therefore staying inside for a long duration of time[15]. Because of their very long habitation time and acidic behaviour, they cause serious harm to the lung tissue (oedema) [16].



Fig 5: SO_2 Conc of four different sampling sites in the month of December 14 – March 15



Fig 6: NO₂ Conc of four different sampling sites in the month of December 14 – March 15.

Hysplit: These climatic conditions i.e. less dispersion and low mixing heights or lower boundary layer height, typically 500m [17] during winter months help the ambient particles to remain for longer time in the atmosphere. These stagnant meteorological conditions is also supported by the results of back trajectory analysis that shows that the site is under the influence of different local emissions and account for increased levels of particulate mass. The effect of localized sources during the months of December and March can be seen in fig. 7.



APPLICATIONS

The analysis of analysis of water soluble ions (WSI) in $PM_{2.5}$ has been done to examine the effect of these ions in the atmosphere. A NOOA HYSPLIT software technique has been applied for the dispersion and mixing heights or lower boundary layer height of air pollutant.

CONCLUSIONS

The analysis of analysis of water soluble ions (WSI) in $PM_{2.5}$ has been done to examine the effect of these ions in the atmosphere. A NOOA HYSPLIT software technique has been applied for the dispersion and mixing heights or lower boundary layer height of air pollutant.

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REFERENCES

- [1] F. Laden, L.M. Neas, D.W. Dockery, J. Schwartz, Association of fine particulate matter from different sources with daily mortality in six U.S. cities, *Environ Health Perspect*, **2000**, 941–947.
- [2] V. Ramanathan, P.J. Crutzen, J.T. Kiehl, D. Rosenfeld, The Indian Ocean Experiment: An Integrated Assessment of the Climate Forcing and Effects of the Great Indo-Asian Haze, *J. Geophys. Res.*, **2001**, 28, 371-398.
- [3] Y.I. Tsai, S.C. Kuo, PM_{2.5} aerosol water content and chemical composition in a metropolitan and a coastal area in southern Taiwan, *Atmos. Environ*, **2005**, 4827–4839.
- [4] R.L. Mariani, W.Z. De Mello, PM_{2.5-10} and associated water-soluble inorganic species at a coastal urban site in the metropolitan region of Rio de Janeiro, *Atmos Environ*, **2007**, 2887-2892.
- [5] S. Jaison, N. Sidhardhan, N. Saxena, A. Mishra, Comparative Study of Heavy Metal Contamination in Sediments of Yamuna river, Agra, *J. Applicable. Chem.*, **2013**, 2 (3), 523-538.
- [6] W.C. Malm, J.F. Sisler, D. Huffman, R.A. Eldred, T.A. Cahill, Spatial and seasonal trends in particle concentration and optical extinction in the United States, *J. Geophys. Res.*, **1994**, 1347-1370.
- [7] K. Karar, A.K. Gupta, A. Kumar, A.K. Biswas, Seasonal Variations of PM₁₀ and TSP in Residential and Industrial Sites in an Urban Area of Kolkata, India, *Environ Monitoring* and Assessment, 2006, 369-381.
- [8] S.N. Behera, M. Sharma, Investigating the Potential Role of Ammonia in Ion Chemistry of Fine Particulate Matter Formation for an Urban Environment, *Sci. Total Environ*, **2010**, 3569–3575.
- [9] D.K. Deshmukh, M.K. Deb, P.K. Hopke, Y. I. Tsai, Seasonal characteristics of water-soluble dicarboxylates associated with PM₁₀ in the urban atmosphere of Durg City, India, *Aerosol and Air Quality Research*, 2012b, 683–696.
- [10] A. Kulshrestha, P.G. Satsangi, J. Masih, A. Taneja, Metal concentration of PM_{2.5} and PM₁₀ particles and seasonal variations in urban and rural environment of Agra, *India, Sci. Total Environ*, 2009, 6196-6204.
- [11] P. Pandey, A.H. Khan, A.K. Verma, K.A. Singh, N. Mathur, G.C. Kisku, S.C. Barman, Seasonal treands of PM_{2.5} and PM₁₀ in ambient air and their correlation in ambient air of Lucknow city, *India, Bull. Environ. Contam. Toxicol*, **2012**, 264-270.
- [12] P. Kothai, I.V. Saradhi, G.G. Pandit, A. Markwitz, V.D. Puranik, Chemical Characterization and Source Identification of Particulate Matter at an Urban Site of Navi Mumbai, India, *Aerosol Air Qual. Res*, 2011, 560–569.
- [13] R. Rengarajan, A.K. Sudheer, M.M. Sarin, Wintertime PM_{2.5} and PM₁₀ carbonaceous and inorganic constituents from urban site in western India, *Atmospheric Research*, 2011b, 420–431.
- [14] M.K. Khan, S.R. Gilani, A. Rauf, K. Shahid, Y. Ali, Estimation of NO_x, SO_x, CO and Particulate Matter from stack emission and its hazardous effect on human health, *J. Applicable. Chem.*, 2015, 4 (1), 204-211.
- [15] B. Ambade, S. Ghosh, B. Shubhankar, Ambient air quality of Jamshedpur City: A study with reference to SO₂, NO₂, RSPM and SPM contents, *J. Applicable. Chem.*, **2013**, 2 (3), 539-544.
- [16] World Health Organization Regional Office for Europe, "Air Quality Guidelines. Global update 2005, Particulate Matter, Ozone, Nitrogen Dioxide and Sulphur Dioxide". http://www.euro.who.int/en/what-we-do/health-topics/environment-and-health/air-quality/publicati ons/pre2009/air-quality-guidelines-global-update-2005,-particulate-matter,-ozone,nitrogen dioxide and sulphur dioxide, 2006, accessed in January 2015.

- [17] V.S. Nair, K.K. Moorthy, D.P. Alappattu, Winter time aerosol characteristics over the Indo-Gangetic Plain (IGP): Impacts of local boundary layer processes and long-range transport, *J. Geophys. Res.*, **2007**, 112-122.
- [18] W. Wang, B.T. Anderson, N. Phillips, R.K. Kaufmann, C. Potter, R. B. Myneni, Feedbacks of vegetation on summertime climate variability over the North American Grasslands. Part I: Statistical analysis. *Earth Interactions*, 2006, 1-30.
- [19] Fang. Xiaomin, Han. Yongxiang, Kang. Shichang, Wang. Huijun, Kang. Fenqing, Long-term monitoring and source apportionment of PM_{2.5}/PM₁₀in Beijing, *China*, J. environ sci., 2008, 1323– 1327.
- [20] K. Ram, M.M. Sarin, Spatio-temporal Variability in Atmospheric Abundances of EC, OC and WSOC over Northern India, *J. Aerosol Sci*, **2010**, 88–98.
- [21] N. Kumar, A. Chu, A. Foster, An empirical relationship between PM_{2.5} and aerosol optical depth in Delhi Metropolitan, *Atmos Environ*, **2007**, 4492–4503.
- [22] S. Tiwari, A.K. Srivastava, D.S. Bisht, T. Bano, S. Singh, S. Behura, M.K. Srivastava, D.M. Chate, B. Padmanabhamurty, Black Carbon and Chemical Characteristics of PM₁₀ and PM_{2.5} at an Urban Site of North India, *J. Atmos. Chem*, **2010**, 193–209.
- [23] N. Verma, S.P. Bagare, S.S. Ningombam, R.B. Singh, Aerosol optical properties retrieved using sky radiometer at Hanle in western Himalayas, *J. Atmos. Sol. Terr. Phys.*, **2009**, 115–124.

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